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(56) Documents Cited

Macromolecules Vol. 30 (7) pp.1890-1896 (1997)  
Macromolecules Vol. 25 (21) pp.5561-5573 (1992)

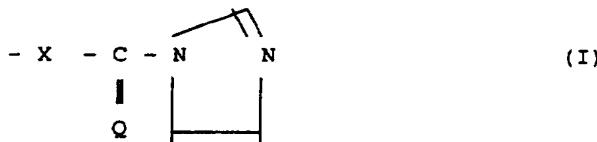
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(54) Abstract Title

**Hyperbranched polymers**

(57) Hyperbranched polymers containing linkages selected from urea, urethane, carbonate, ester and amine linkages are described. A hyperbranched polymer is prepared by reacting a polyfunctional compound having three reactive functional groups (B) in substantially equimolar quantities with a reagent which reacts rapidly with group (B) to form a group (A) which reacts more slowly with group (B), or by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reactive with (A), either (A) or (B) being a group of the formula:



where Q represents O or S and X represents -O-, -S-, -NH- or a direct bond.

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Polymers and their Preparation

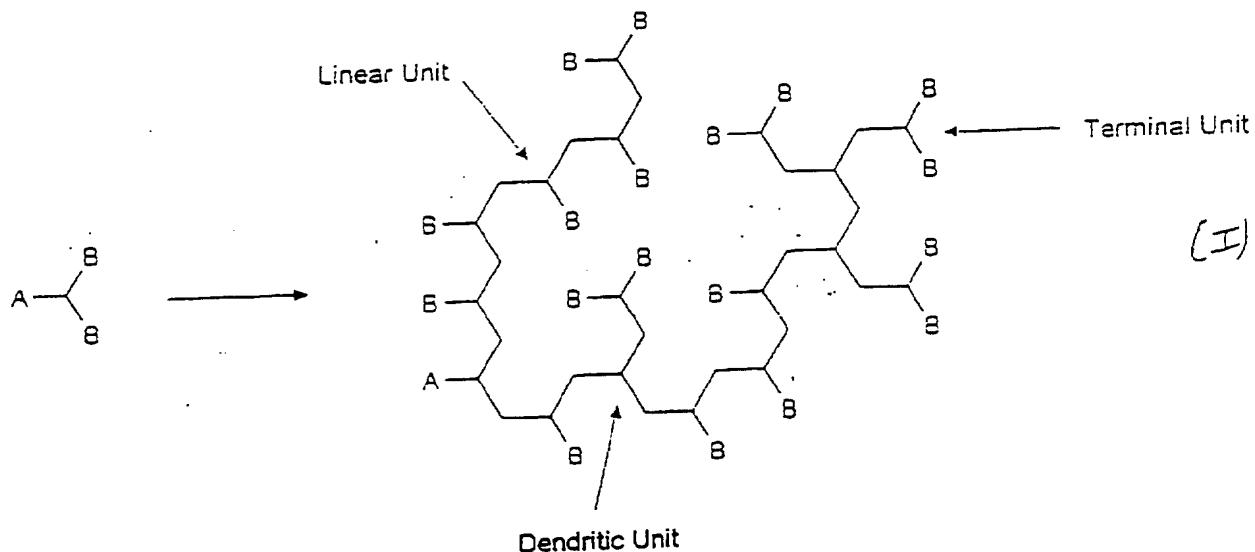
This invention relates to highly branched polymers known as hyperbranched polymers and to reactions for their preparation.

Hyperbranched polymers are generally prepared by the polymerisation in a single reaction of a compound (monomer) containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reactive with (A). Such a monomer will react as shown in Figure I below to form a branched polymer containing one group (A) and  $(x+1)$  groups (B) if  $x$  monomer molecules have reacted together.

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The polymerisation of an  $AB_2$  monomer leads to a complex mixture of products containing both linear and dendritic regions. 3 subunits are found within a hyperbranched polymer formed from an  $AB_2$  monomer. These differ in the number of B

functionalities which have reacted. If both have reacted a dendritic unit is formed, if only one has reacted a linear unit is formed and if no B groups have reacted a terminal unit is formed.

5

The ratio of these units can be used to determine the degree of branching of the polymer.

Degree of Branching = Number of Dendritic Units + Number of Terminal Units

10

Total number of Units

The degree of branching for an ideal dendrimer is equal to 1. A linear polymer has a degree of branching of 0. A hyperbranched polymer has a structure intermediate between a 15 dendrimer and linear polymer and can in theory have a degree of branching between 0 and 1 but generally is only considered as a hyperbranched polymer when the degree of branching is greater than 0.4, as described in US-A-5514764.

20 Hyperbranched polyesters are described for example in US-A-5418301, US-A-5514764 and WO-A-96/12754. Hyperbranched polycarbonates are described by D.H. Bolton and K. L. Wooley in *Macromolecules* 1997, vol. 30, page 1890. Hyperbranched polyphenylenes are described by Y.H. Kim and O.W. Webster in 25 *Macromolecules* 1992, 25, 5562. Hyperbranched polyethers are described by K. E. Uhrich, C.J. Hawker, J.M.J. Frechet and S.R. Turner in *Macromolecules*, 1992, 25, 4583. Hyperbranched polyetherketones are described by F. Chu and C.J. Hawker in *Polymer Bulletin*, 1993, 30, 265. Hyperbranched polyurethanes 30 are described by A. Kumar and S. Ramakrishnan in *J. Chem. Soc. Commun.* 1993, 18, 1453 and by R. Spindler and J.M.J. Frechet in *Macromolecules*, 1993, 26, 4809. Hyperbranched

poly(ester-amide)s are described by H.R. Kricheldorf and G. Lohden in *Macromol. Chem. Phys.*, 1995, 196, 1839.

Hyperbranched polymers have wide industrial potential  
5 for use in high solids coatings and sealants, as reactive  
plasticisers and as substrates for binding biological  
reagents such as enzymes or antigens owing to their  
characteristics such as lower viscosity when molten or in  
solution when compared to linear analogues, higher solubility  
10 than their linear analogues and high chemical reactivity  
(large number of reactive chain ends). There is a need for  
further reactions capable of producing hyperbranched polymers  
without premature crosslinking and without needing protection  
and deprotection of reactive groups.

15

A process according to one aspect of the present invention for the preparation of a hyperbranched polymer by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a  
20 different chemical type which is reactive with (A) is characterised in that the single group (A) is a group of the formula:

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where Q represents O or S and X represents -O-, -S-, -NH- or a direct bond, the group being linked to the remainder of the compound or polymer through a carbon atom.

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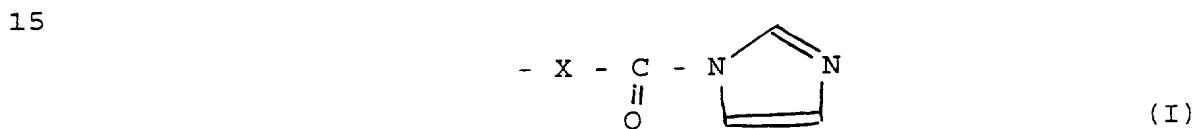
According to another aspect of the invention, a process for the preparation of a hyperbranched polymer by the

polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reactive with (A), characterised in that either the group (A) or the group (B) is a group of the 5 formula:-



10 where Q represents O or S and X' represents -NH- or a direct bond, the group being linked to the remainder of the compound through a carbon atom.

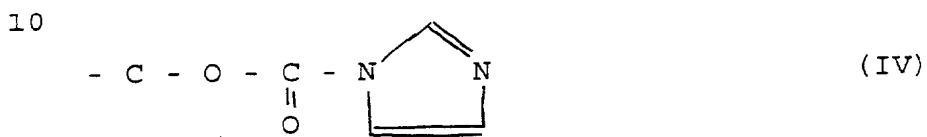
The imidazolide group of the formula:



can in general be produced by reaction of a compound or 20 polymer containing hydroxyl, thiol, amine or carboxylic acid groups with carbonyl diimidazole or thiocarbonyl diimidazole. The imidazole ring can be substituted or can be fused to a benzene ring, leading to a substituted imidazole ring in the group of formula (I), but this is not preferred. The 25 preferred reagent is carbonyl diimidazole. Alcohol groups, thiol groups, primary amino groups and carboxylic acid groups react with carbonyl diimidazole to form an imidazolide in which an active hydrogen atom of the alcohol, thiol and/or amino groups, and/or the whole carboxylic acid group is 30 replaced by a carbonyl imidazole group of the formula:

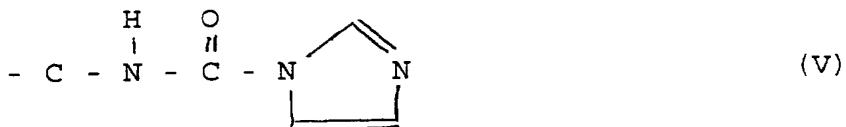


5 The term "imidazolide" is used in this specification to  
describe a 1-imidazole carboxyl ester of an alcohol having  
the structure:



or a 1-imidazole carboxamide having the structure:

15



as well as an imidazolide having the structure:

20



The imidazolide group can be prepared in alternative ways, although preparation from carbonyl diimidazole is most convenient. 1-Imidazole carboxyl esters can be prepared by reaction of a chloroformate with imidazole or by reaction of the carbonyl chloride of imidazole with an alcohol. 1-Imidazole carboxamides can be prepared by reaction of a carbamoyl chloride with imidazole or by reaction of the carbonyl chloride of the imidazole with an amine. 1-Imidazole

carbonyl compounds can be prepared by reaction of an acyl chloride with imidazole.

The imidazolide group can in general be the single A 5 group in an  $AB_2$  monomer, forming on polymerisation a hyperbranched polymer having terminal groups B which are the functional groups reactive with the imidazolide, or the groups B in an  $AB_2$  monomer can be imidazolide groups, forming a hyperbranched polymer having terminal imidazolide groups. 10 These terminal imidazolide groups can be reacted in a further step to convert them to different groups if desired.

The group reacted with carbonyl diimidazole to form the group (A) or (B) of an  $AB_2$  monomer is preferably a carboxylic 15 acid group or primary amine group. Carboxylic acid groups react with carbonyl diimidazole to replace one imidazole ring to form an imidazolide readily at ambient temperature without need for a catalyst with little tendency towards further reaction to replace the remaining imidazole ring. Primary 20 amines are capable of reacting with carbonyl diimidazole at room temperature without need for a catalyst to replace both imidazole rings, forming a urea linkage, if the primary amine is present in stoichiometric excess, but if the reagents are used in stoichiometrically equivalent amounts or if the 25 carbonyl diimidazole is used in excess, the predominant reaction is to replace only one of the imidazole rings of the carbonyl diimidazole.

According to another aspect of the invention, a process 30 for the preparation of a hyperbranched polymer by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a different

chemical type which is reactive with (A) is characterised in that a precursor compound, containing either a single group (A') of one chemical type reactive with carbonyl diimidazole and at least two groups (B) of a different chemical type from 5 A' which is less reactive than A' with carbonyl diimidazole or at least two groups (B') of one chemical type reactive with carbonyl diimidazole and a single group (A) of a different chemical type from B' which is less reactive than B' with carbonyl diimidazole, is reacted with carbonyl 10 diimidazole under conditions such that only the group A' or the groups B' react with the carbonyl diimidazole to form imidazolide groups, and the resulting compound containing imidazolide groups is polymerised under more vigorous reaction conditions to form the hyperbranched polymer.

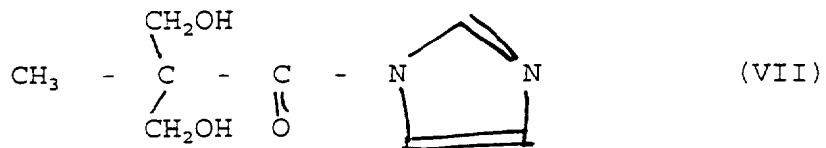
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Alcohol groups will react readily with carbonyl diimidazole in the presence of a basic catalyst but only very slowly at room temperature without catalyst. Primary alcohol groups can react to replace both imidazole rings, forming a 20 carbonate linkage, and are preferably not used in stoichiometric excess. Secondary and tertiary alcohol groups generally show little tendency to further reaction to replace the remaining imidazole ring.

25 This requirement for a catalyst or for heating can be used in a convenient method for preparing an  $AB_2$  monomer and then polymerising it to form a hyperbranched polymer according to the invention. For example a precursor compound containing one carboxylic acid group (A') and two hydroxyl, 30 groups such as 2,2-bis(hydroxymethyl)propionic acid or 3,5-dihydroxybenzoic acid or the N,N-bis(2-hydroxyethyl) monoamide of adipic acid can be reacted with carbonyl

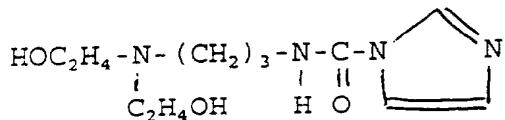
diimidazole at room temperature without catalyst to form an imidazolide such as

5



The hydroxyl group can for example be a primary alcohol group or an unhindered secondary alcohol group such as a -  
10 CHOHCH<sub>3</sub> group or a phenol group when the carboxylic acid is aliphatic; almost all primary and secondary alcohol and phenol groups are suitable if the carboxylic acid is aromatic. This compound containing one imidazolide group and two alcohol groups can be used as an AB<sub>2</sub> monomer. It is substantially stable at ambient temperature (up to 30°C) in  
15 the absence of catalyst. Under more vigorous conditions, for example when heated to at least 50°C, preferably 50-120°C, intermolecular reaction takes place between an alcohol group and an imidazolide group, forming a larger molecule containing one imidazolide group together with an ester  
20 linkage and alcohol groups and eventually a hyperbranched polyester having terminal hydroxyl groups. Alternatively the imidazolide can be polymerised by addition of a basic catalyst, for example KOH, NaOH, a quaternary ammonium hydroxide or a strongly basic amine such as  
25 diazabicycloundecene.

Similarly a precursor compound containing one primary amine group (A') and two primary or secondary alcohol groups such as N-(3-aminopropyl) diethanolamine can be reacted with  
30 a substantially equimolar amount of carbonyl diimidazole to form an imidazolide such as



(VIII)

This is stable at ambient temperature in the absence of catalyst and can be used as an  $AB_2$  monomer which is activated by heating or by basic catalyst, forming a hyperbranched polyurethane having terminal hydroxyl groups.

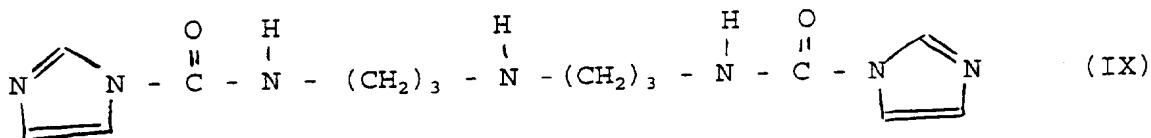
Alternatively the precursor compound can contain two or more carboxylic groups (B') and a single hydroxyl group, for example 5-hydroxyisophthalic acid, or two or more primary amine groups (B') and a single primary or secondary alcohol group, for example N,N'-bis(3-aminopropyl)ethanolamine. Such a precursor compound can be reacted with carbonyl diimidazole to produce a compound containing a single hydroxyl group and two or more imidazolide groups. In the case of the diamine, a substantially stoichiometric amount of carbonyl diimidazole (2 moles per mole diamine) must be used. This compound can be polymerised by heating or addition of basic catalyst to form a hyperbranched polymer having terminal imidazolide groups.

It is generally preferred that compounds having functional groups on adjacent carbon atoms, such as glycerol or 2-amino-1,3-propanediol or diethylene triamine are not reacted with carbonyl diimidazole in the process of the invention, since such compounds tend to undergo a cyclisation reaction forming a cyclic carbonate, urethane or urea which is not useful in preparing hyperbranched polymers.

30

Hyperbranched polyureas can be formed by the reaction of a diprimary secondary triamine as precursor compound in which

the primary amine groups act as reactive groups B', for example dipropylene triamine, bis(hexamethylene) triamine or N-(2-aminoethyl)-propane-1,3-diamine, with carbonyl diimidazole at low temperature, preferably 0°C or below 5 followed by heating to ambient temperature (e.g. 20 or 25°C) or above to polymerise the resulting AB<sub>2</sub> monomer. Secondary amine groups react with carbonyl diimidazole at room temperature or above but the imidazolide formed is rather stable and does not readily react with any alcohol or amine 10 groups. Secondary amine groups react readily with imidazolide groups formed from primary amines, aromatic acids or primary alcohols. Primary amine groups react more rapidly with carbonyl diimidazole than do secondary amine groups. Thus if dipropylene triamine is reacted with carbonyl diimidazole at 15 0°C the preponderant product is



20 When this is polymerised, the hyperbranched polyurea product has a plurality of terminal imidazolide groups. Alternatively a primary disecondary amine can be reacted with carbonyl diimidazole to produce a compound containing one imidazolide group and two secondary amine groups and this can be 25 polymerised at room temperature or above to produce a hyperbranched polyurea.

30 Imidazolide groups are generally reactive with amine groups and with most primary or secondary hydroxyl groups; a hyperbranched polymer having terminal imidazolide groups can thus be used as a crosslinking agent in coating or sealant compositions. Alternatively, such a hyperbranched polymer can

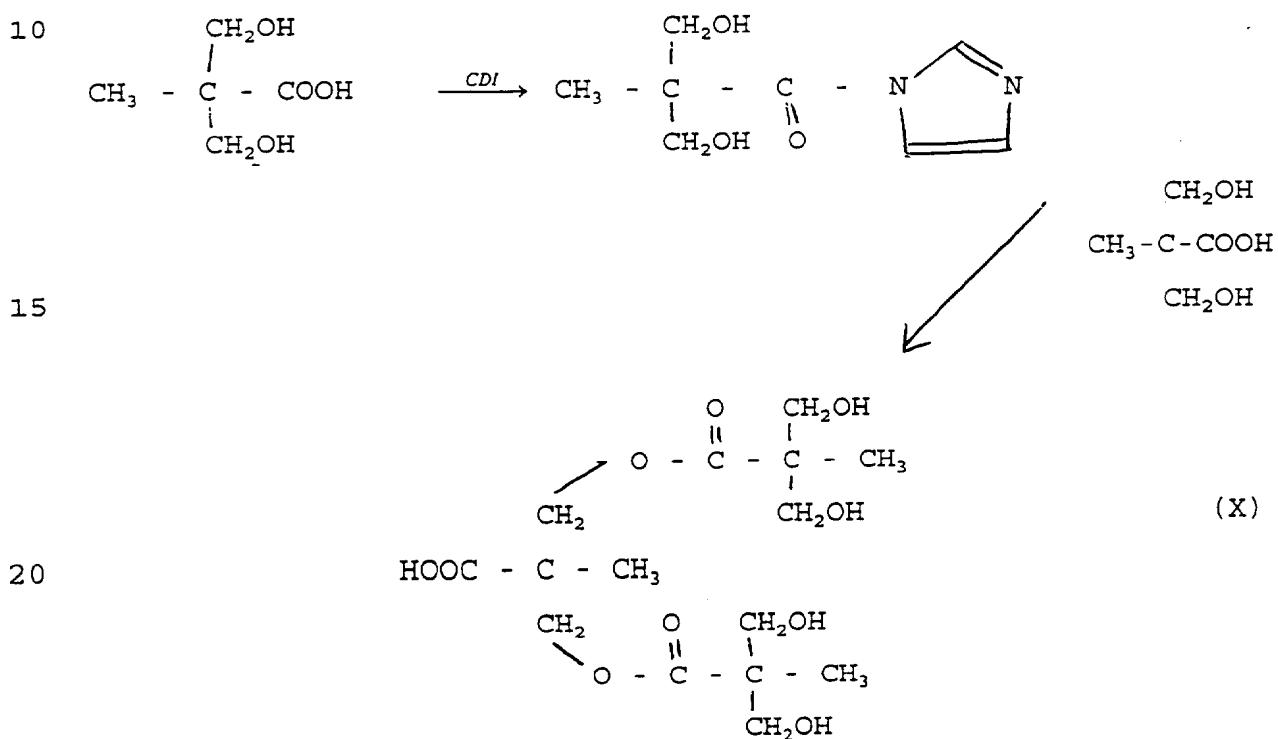
be reacted with a monofunctional amine or primary or secondary alcohol to form a non-functional hyperbranched polymer, or can be reacted with an amine or primary or secondary alcohol containing a functional group (C) not reactive with the imidazolide group, for example a carboxylic acid group to produce a hyperbranched polymer having functionality (C). Mixtures of reagents, for example a mixture of a monoamine and an aminoacid, can be used to form a hyperbranched polymer having a controlled number of functional groups. Alternatively the hyperbranched polymer can be reacted with water to hydrolyse the imidazolide group. For example the terminal imidazolide groups of the hyperbranched polyurea described above can be hydrolysed to form terminal amine groups. Terminal imidazolide groups derived from a carboxylic acid hydrolyse very readily to form terminal carboxylic acid groups.

Any of the above processes can be modified to increase or decrease the degree of branching of the hyperbranched polymer. The degree of branching can be decreased by the incorporation of an AB monomer, which can only form linear units, with the AB<sub>2</sub> monomer. For example a mixture of a dihydroxycarboxylic acid and a monohydroxycarboxylic acid such as lactic acid, gamma-hydroxybutyric acid or 2,2-dimethyl-3-hydroxypropionic acid can be reacted with carbonyl diimidazole to produce a mixture of imidazolides which act as AB<sub>2</sub> and AB monomers and can be polymerised by heating or by strong base catalyst to form a branched polyester containing an increased proportion of linear units.

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The degree of branching can be increased by the use of an AB<sub>4</sub> compound having a dendritic structure as all or part of

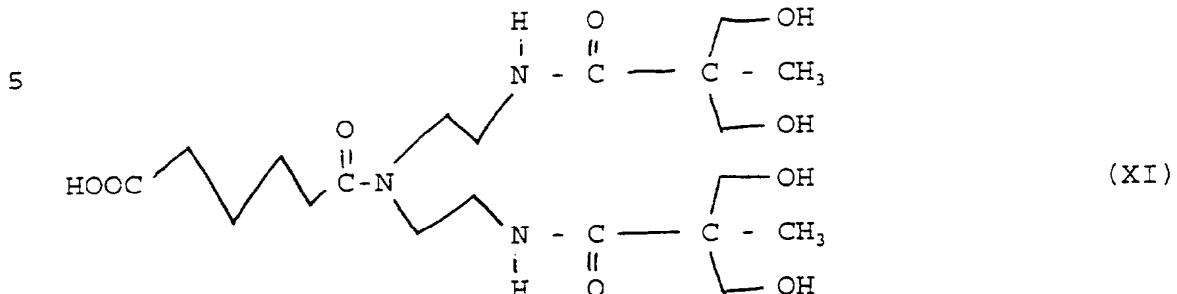
the monomer forming the hyperbranched polymer. For example a dihydroxycarboxylic acid such as bis(hydroxymethyl)propionic acid can be reacted with carbonyl diimidazole as described above and the resulting imidazolide can be reacted with further dihydroxycarboxylic acid by heating or in the presence of strong base catalyst to form a dendritic compound having one carboxylic acid group and four hydroxyl groups as shown below.



25 This can be reacted with carbonyl diimidazole to form an  
AB<sub>4</sub> monomer which can be polymerised to a hyperbranched  
polymer by heating or by a strong base catalyst.

Alternatively the imidazolide of a dihydroxycarboxylic acid can be reacted with a diaminocarboxylic acid such as an N,N-bis(2-aminoethyl) monoamide of a dicarboxylic acid to

prepare a dendritic compound containing one carboxylic acid group and four hydroxyl groups.



10

This can be reacted with carbonyl diimidazole to form an  $AB_4$  monomer and polymerised by heating or by strong base catalyst to form a hyperbranched polymer containing both amide and ester linkages.

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An alternative method according to the invention for the preparation of hyperbranched copolymers is by the copolymerisation of two or more different imidazolide compounds. Hyperbranched copolymers containing different linkages can be prepared by copolymerising imidazolides in which the groups of formula (I) are different. For example the imidazolide of a dihydroxycarboxylic acid can be copolymerised with the imidazolide of a dihydroxyamine (in which the hydroxyl groups are primary or secondary alcohol groups and the amine group is a primary or secondary amine group) such as N-(3-aminopropyl) diethanolamine to form a hyperbranched copolymer containing both ester and urethane links.

30 Thus, according to another aspect of the invention a hyperbranched polymer formed by the polymerisation of at least one compound containing a single group (A) of one

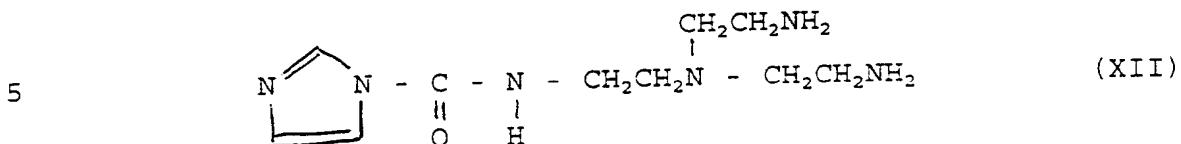
chemical type and at least two groups (B) of a different chemical type which is reacted with (A) is characterised in that the hyperbranched polymer is a copolymer containing two different chemical linkages, one of said linkages being selected from urea, urethane and carbonate linkages and the other of said linkages being selected from urea, urethane, carbonate, ester and amide linkages.

Alternatively, hyperbranched copolymers containing both aromatic and aliphatic groups can be produced by copolymerising two imidazolides, one of which is an aromatic compound such as the imidazolide of 3,5-dihydroxybenzoic acid and the other of which is an aliphatic compound such as the imidazolide of 2,2-bis(hydroxymethyl) propionic acid.

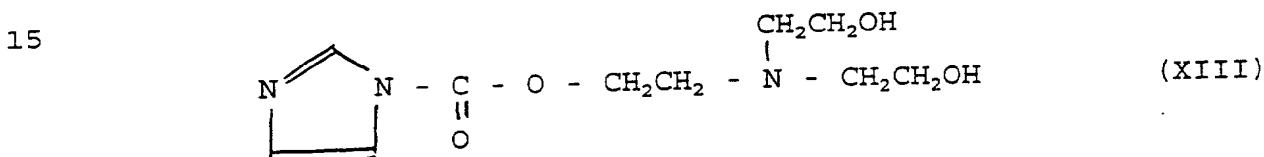
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A process according to another aspect of the invention for the preparation of a hyperbranched polymer is characterised in that a polyfunctional compound having at least three reactive functional groups (B) is reacted in substantially equimolar quantities with a reagent which reacts rapidly with group (B) to form a group (A) which reacts more slowly with group (B). Examples of such reagents are carbonyl diimidazole and thiocarbonyl diimidazole, which react rapidly with primary amino groups or primary alcohol groups to form imidazolide groups which react less rapidly with the remaining primary amino or primary alcohol groups. The reaction with primary amine can be carried out at ambient temperature in the absence of catalyst. A strong base catalyst or elevated temperature, for example 50 to 120°C, is needed with the alcohol. An example of a suitable tri(primary amine) is tris(2-aminoethyl)amine and an example of a suitable tri(primary alcohol) is triethanolamine. The

carbonyl diimidazole reacts rapidly with one primary amine group, predominantly forming the  $AB_2$  monomer



before the subsequent slower reaction of the imidazolide group with the primary amine groups to form urea linkages and eventually an amine-functional hyperbranched polyurea. We believe that hyperbranched polyureas have not previously been prepared. Similarly triethanolamine reacts rapidly to form predominantly



before the subsequent slower reaction to form carbonate linkages and eventually a hydroxy-functional hyperbranched polycarbonate.

The reagent which reacts rapidly with group (B) can alternatively be a compound containing two imidazolide groups of formula (I), for example an imidazolide of a dicarboxylic acid such as adipic or sebacic acid or an imidazolide of a di(secondry or tertiary alcohol) such as hexane-2,5-diol or 2,5-dimethylhexane-2,5-diol or an imidazolide of a di(primary amine) such as hexane-1,6-diamine. The imidazolide of a dicarboxylic acid can be reacted with a primary triamine such as tris(2-aminoethyl)amine to produce a hyperbranched polyamide or with a triol such as triethanolamine to produce

a hyperbranched polyester. The imidazolide of a diol can be reacted with a triol such as triethanolamine to produce a hyperbranched polycarbonate or with a primary triamine such as tris(2-aminoethyl) amine to produce a hyperbranched 5 polyurethane. The imidazolide of a diamine can be reacted with a primary triamine to produce a hyperbranched polyurea or with a triol to produce a hyperbranched polyurethane.

Alternatively, a compound containing three imidazolide 10 groups derived from a tri-carboxylic acid or a triol such as tris(2-hydroxypropyl)amine or a triamine such as tris(2-aminoethyl)amine can act as the polyfunctional compound having at least three functional groups (B), with for example 15 a primary diamine acting as the reagent which reacts rapidly with group(B). The imidazolide of a tricarboxylic acid can be reacted with a primary diamine such as ethylene diamine to produce a hyperbranched polyamide or with a diol such as pentane-1,5-diol to produce a hyperbranched polyester. The imidazolide of a triol can be reacted with a primary diamine 20 to produce a hyperbranched polyurethane or with a diol to produce a hyperbranched polycarbonate. The imidazolide of a triamine can be reacted with a diamine to produce a hyperbranched polyurea or with a diol to produce a hyperbranched polyurethane.

25

Many of the hyperbranched polymers of the invention, have reactive functional groups at the surface and are thereby suitable for use as ingredients of high solids 30 coatings and sealants; they can be cured through their functional groups to form solid coating films. They can also be used as reactive plasticisers in thermoplastic compositions. The hyperbranched polymers can alternatively be

used as rheology modifiers in coating and sealant compositions, causing a greater reduction of viscosity than a similar amount of volatile solvent.

5 Hyperbranched polymers having reactive functional groups at the surface can be used to bind biological reagents such as enzymes or antigens, as described for example in US-A-5229490, providing a high concentration of the biological reagent in a low molecular volume. Alternatively, new 10 colourants can be formed by attaching chromophores at the surface of the hyperbranched polymer.

In some uses it may be preferred to remove or reduce the 15 number of surface functional groups. Hydroxyl groups can for example be esterified. A hydroxy-functional hyperbranched polyurethane which is water-soluble is rendered water-insoluble if most of its hydroxyl groups are esterified to benzoate groups. Similarly amine functional groups can be converted to amide groups to reduce water solubility. The 20 surface functional groups can if desired be reacted with a compound containing a reactive functional group and one or more non-reactive groups which affect the properties of the polymer, for example fluoro or silicone groups to give low surface energy.

25 It will be seen from Figure 1 that the hyperbranched polymer contains one group A. It is assumed that this has become unreactive with groups B due to steric effect. This group A can be reacted with a compound having two or more 30 groups more reactive than group B to couple two or more hyperbranched polymer molecules. For example a hyperbranched polyester containing an imidazolide group derived from a

dihydroxycarboxylic acid can be coupled by a di(primary amine).

The invention is illustrated by the following Examples:-

5

Example 1

A solution of N-(3-aminopropyl)-diethanolamine (3.00g, 18.5 mmol) in tetrahydrofuran (THF) (30 ml) was added 10 dropwise to a solution of carbonyl diimidazole (3.00g, 18.5 mmol) in THF (45 ml) and stirred at room temperature for 20 hours to form an imidazolide.

Potassium hydroxide (0.1g, 1.8 mmol) was added to the 15 reaction mixture and the solvent removed in vacuo. The residue was then heated at 120°C for 3 hours. THF (60 ml) was added to the residue and heated at 60°C for 1 hour. The THF was decanted and the residue dried in vacuo to give a yellow 20 gum. This was a hyperbranched urethane polymer of maximum molecular weight 6000 and number average molecular weight Mn 1075 as measured by MALDI (matrix assisted laser desorption ionisation mass spectrometry). The degree of branching was estimated at 0.6 by <sup>15</sup>N NMR.

25

Example 2

A solution of bishydroxymethylpropionic acid (2.00 g, 15 mmol) and carbonyl diimidazole (2.40g, 15 mmol) in THF (60 ml) were stirred at room temperature for 2 hours.

30

Potassium hydroxide (0.2g, 3.6 mmol) was added to the reaction mixture and the solvent removed in vacuo. The

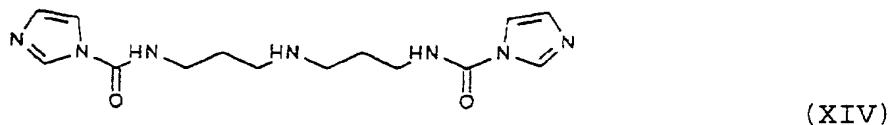
residue was then heated at 90°C for 3 hours. THF (60 ml) was added to the residue and heated at 60°C for 1 hour. The THF was decanted and the residue dried in vacuo to give a colourless gum. This was a hyperbranched polyester of M<sub>w</sub> 2426  
5 and M<sub>n</sub> 2421 as measured by MALDI. The degree of branching was measured as 0.49 by quantitative carbon-13 NMR; the chemical shift of the quaternary carbon atom varies according to whether it is surrounded by 3 ester units (dendritic unit), 2 ester units (linear unit) or one ester unit (terminal unit).

10

Example 3

A solution of dipropylene triamine (2.00g, 15mmol) in THF (10ml) was added to a solution of carbonyl diimidazole  
15 4.95g, 30mmol) in THF (70ml) at 0°C and stirred at 0°C for 2 hours to form an AB<sub>2</sub> monomer of the formula:

20



Without isolation, the product was then reacted at room  
25 temperature for 20 hours. The reaction mixture was concentrated in vacuo to give a hyperbranched polyurea as a white sticky solid. Molecular weight was measured by MALDI and gave maximum molecular weights up to 3500.

30

Example 4

Carbonyl diimidazole (2.00g, 12mmol) was added to a

solution of trisaminoethylamine (1.80g, 12mmol) in THF (100ml) at 0°C and stirred for 4 hours at 0°C and then at room temperature for 18 hours. The THF was decanted, the residue dissolved in methanol and then concentrated in vacuo 5 to give a hyperbranched polyurea as a yellow liquid. Molecular weight was measured by MALDI and gave maximums up to 3000.

10

Example 5

Carbonyl diimidazole (3.63g, 22mmol), trimethylolpropane (3.00g, 22mmol) and potassium hydroxide (0.2g, 3.5mmol) in THF (50ml) were heated at 60°C for 3 hours. The solvent was 15 removed and the residue heated at 60°C for a further 3 hours to give a hyperbranched polycarbonate as a colourless gum. Molecular weight was measured by MALDI and gave maximums up to 5000.

20

Example 6

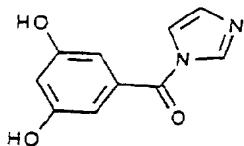
Carbonyl diimidazole (3.00g, 19mmol) was added to sebacic acid (1.87g, 9.3mmol) in methanol (50ml) and stirred 25 at room temperature for 2 hours to form the bis(imidazclide) of sebacic acid. Trisaminoethylamine (1.35g, 9.3mmol) was added to the solution and the reaction stirred at room temperature for 96 hours then concentrated in vacuo to give a hyperbranched polyamide as a yellow liquid.

30

Example 7

A solution of 3,5-dihydroxybenzoic acid (2.00g, 13mmol) and carbonyl diimidazole (2.1g, 13mmol) in THF (60ml) was  
5 stirred at room temperature for 2 hours to form an imidazolide of the formula:

10



(XV)

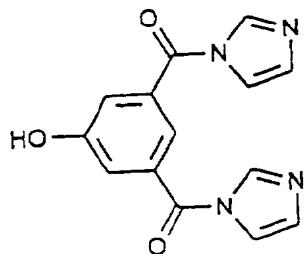
20

Potassium hydroxide (0.2g, 3.6mmol) was added to the solution  
15 and the reaction mixture stirred at room temperature for 24 hours then concentrated in vacuo. Water was added to the residue and the pale brown precipitate filtered and dried in vacuo to give a hydroxy-terminated hyperbranched polyester as a brown solid. Molecular weight was measured by MALDI and  
20 gave maximums up to 3500.

Example 8

A solution of 5-hydroxyisophthalic acid (1.69g, 9.3mmol) and carbonyl diimidazole (3.00g, 19mmol) in THF (60ml) was  
25 stirred at room temperature for 2 hours to form a bis(imidazolide) of the formula:

30



(XVI)

The solvent was removed and the residue heated at 90°C for 3 hours and cooled. THF was added to the residue, the white precipitate was removed by filtration and dried in vacuo to give a hyperbranched polyester as a white solid. Molecular weight was measured by MALDI and gave maximums up to 6000. The hyperbranched polyester had terminal acid functionality as a result of hydrolysis of imidazolide groups by ambient moisture during work-up.

10

Example 9 - Synthesis of a Hyperbranched Polyester/Polyurethane Copolymer

Step 1: A solution of 3-aminopropylidethanolamine (3.00g, 19mmol) in THF (30ml) was added to a solution of carbonyl diimidazole (3.00g, 19mmol) in THF (45ml) and stirred at room temperature for 20 hours to form imidazolide (VIII).

Step 2: A solution of bishydroxymethylpropionic acid (2.48g, 19mmol) and carbonyl diimidazole (3.00g, 19mmol) in THF (60ml) was stirred at room temperature for 2 hours to form imidazolide (VII).

Step 3: The reaction mixtures produced in steps 1 and 2 were combined and concentrated in vacuo. The residue was heated at 90°C for 3 hours and cooled. THF (100ml) was added to the residue and heated at 60°C for 1 hour. The THF was decanted and the residue dried in vacuo to give the copolymer as an orange gum.

Example 10 - Synthesis of a Hyperbranched Aliphatic Polyester/Aromatic Polyester Copolymer

Step 1: A solution of 3,5-dihydroxybenzoic acid (1.90g, 5 12mmol) and carbonyl diimidazole (2.00g, 12mmol) in THF (60ml) was stirred at room temperature for 2 hours to form imidazolide (XV).

Step 2: A solution of bishydroxymethylpropionic acid 10 (1.65g, 12mmol) and carbonyl diimidazole (2.00g, 12mmol) in THF (60ml) was stirred at room temperature for 2 hours to form (VII).

Step 3: The reaction mixtures produced in steps 1 and 2 15 were combined and concentrated in vacuo. The residue was heated at 90°C for 3 hours and cooled. THF (100ml) was added to the residue and heated at 60°C for 1 hour. The THF was decanted and the residue dried in vacuo to give the copolymer as a pale brown solid.

20

Example 11 - Modification of the Surface Functionality of a Hyperbranched Polyurethane

25 A solution of benzoic acid (0.57g, 4.7mmol) and carbonyl diimidazole (0.76g, 4.7mmol) in THF (30ml) were stirred at room temperature for 2 hours to form the imidazolide. The imidazolide so prepared was added to the hydroxy-terminated hyperbranched polyurethane produced in Example 2 (0.76g) and 30 potassium hydroxide (0.1g, 1.7mmol) and the reaction mixture stirred at 60°C for 5 hours. The reaction mixture was

concentrated in vacuo to give the hyperbranched polyurethane terminated with benzonate ester groups. NMR analysis revealed the presence of the benzoate ester groups and the absence of any hydroxyl groups indicating that complete reaction of all 5 of the original hydroxy groups had occurred.

CLAIMS

1. A process for the preparation of a hyperbranched polymer, characterised in that a polyfunctional compound having three reactive functional groups (B) is reacted in 5 substantially equimolar quantities with a reagent which reacts rapidly with group (B) to form a group (A) which reacts more slowly with group (B).
2. A process according to claim 1, characterised in that the reagent is carbonyl diimidazole and the groups (B) are 10 primary amine or primary alcohol groups.
3. A process according to claim 1, characterised in that the reagent is an imidazolide containing two groups of the formula:-

15



where Q represents O or S and X represents -O-, -S-, -NH- or a direct bond, the group being linked to the remainder of the compound through a carbon atom, and the polyfunctional compound is a tris(primary amine) or a tris (primary 20 alcohol).

4. A process according to claim 1, characterised in that the said polyfunctional compound is an imidazolide containing three groups of the formula (I) as defined in claim 4 and the reagent is a compound containing two primary 5 amine or primary alcohol groups.

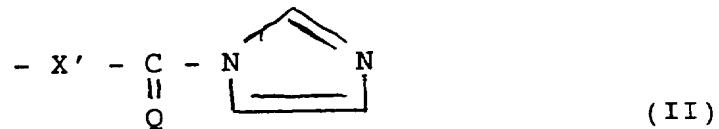
5. A hyperbranched polymer characterised in that all or most of the chemical linkages in the polymer are urea linkages.

6. A process for the preparation of a hyperbranched polymer 10 by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reactive with (A), characterised in that the single group (A) is a group of the formula:

15

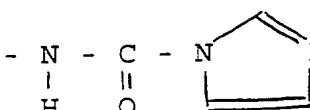


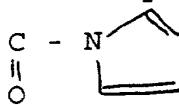
7. A process for the preparation of a hyperbranched polymer by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) 20 of a different chemical type which is reactive with (A), characterised in that either the group (A) or the group (B) is a group of the formula:

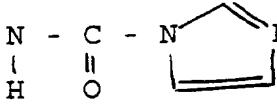


where Q represents O or S and X' represents -NH- or a direct bond, the group being linked to the remainder of the compound through a carbon atom.

5 8. A process according to claim 7 for the preparation of a hyperbranched polyurea, characterised in that the groups (B) are

10  groups and the group (A) is a secondary  
amine group.

9. A process according to claim 7 for the preparation of a hyperbranched polyester, characterised in that the group (II)  
15 is a  group and the other group (A) or (B) is a  
hydroxyl group.

10. A process according to claim 7 for the preparation of a  
20 hyperbranched polyurethane, characterised in that the group  
(II) is a  group and the other  
group (A) or (B) is a primary or secondary alcohol group.

25 11. A process for the preparation of a hyperbranched polymer by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reactive with (A), characterised in that a precursor compound, containing either  
30 a single group (A') of one chemical type reactive with carbonyl diimidazole and at least two groups (B) of a different chemical type from A' which is less reactive than

A' with carbonyl diimidazole or at least two groups (B') of one chemical type reactive with carbonyl diimidazole and a single group (A) of a different chemical type from B' which is less reactive than B' with carbonyl diimidazole, is 5 reacted with carbonyl diimidazole under conditions such that only the group A' or the groups B' react with the carbonyl diimidazole to form imidazolide groups and the resulting compound containing imidazolide groups is polymerised under more vigorous reaction conditions to form the hyperbranched 10 polymer.

12. A process according to claim 11 for the preparation of a hyperbranched polyester, characterised in that the group A' or groups B' of the precursor compound are carboxylic acid 15 groups and the groups B or A respectively of the precursor compound are hydroxyl groups, and the more vigorous reaction conditions comprise heating or catalysis by a basic catalyst.

13. A process according to claim 11 for the preparation of a 20 hyperbranched polyurethane, characterised in that the group A' or groups B' of the precursor compound are primary amine groups and the groups B or A respectively of the precursor compound are primary or secondary alcohol groups, and the more vigorous reaction conditions comprise heating or 25 catalysis by a basic catalyst.

14. A process according to claim 11 for the preparation of a hyperbranched polyurea, characterised in that the precursor compound is a diprimary secondary amine and is reacted with 30 carbonyl diimidazole at 0°C or below, and the more vigorous reaction conditions comprise a higher temperature which is ambient temperature or above.

15. A process for the preparation of a hyperbranched polymer by the polymerisation of a compound containing a single group (A) of one chemical type and at least two groups (B) of a  
5 different chemical type which is reactive with (A), characterised in that either the group (A) or the group (B) is a group of formula (I) as defined in claim 6 and in that a hyperbranched copolymer is formed by polymerising together at least two such compounds.

10 16. A process according to claim 15, characterised in that the groups of formula (I) in one of the said two compounds are different from the groups of formula (I) in the other compound.

15 17. A process according to claim 15, characterised in that one of the said two compounds is an aromatic compound and the other is an aliphatic compound.

20 18. A hyperbranched polymer formed by the polymerisation of at least one compound containing a single group (A) of one chemical type and at least two groups (B) of a different chemical type which is reacted with (A), characterised in that the hyperbranched polymer is a copolymer containing two  
25 different chemical linkages, one of said linkages being selected from urea, urethane and carbonate linkages and the other of said linkages being selected from urea, urethane, carbonate, ester and amine linkages.



The  
Patent  
Office

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**Application No:** GB 9709045.0  
**Claims searched:** 1-4

**Examiner:** Alan Kerry  
**Date of search:** 28 July 1997

**Patents Act 1977**  
**Search Report under Section 17**

**Databases searched:**

UK Patent Office collections, including GB, EP, WO & US patent specifications, in:

UK Cl (Ed.O): C3R RSM, RSP

Int Cl (Ed.6): C08G 63/06, 64/38, 69/24, 71/02, 71/04

Other: Online databases: WPI, CLAIMS

**Documents considered to be relevant:**

Category	Identity of document and relevant passage	Relevant to claims
X	Macromolecules, Vol. 30, No. 7, pp. 1890-1896 (1997)	1
X	Macromolecules, Vol. 25, No. 21, pp. 5561-5573 (1992)	1

X Document indicating lack of novelty or inventive step  
Y Document indicating lack of inventive step if combined with one or more other documents of same category.  
& Member of the same patent family

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